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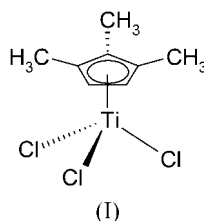
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Key indicators

Single-crystal X-ray study
 $T = 150$ K
Mean $\sigma(\text{C}-\text{C}) = 0.002$ Å
 R factor = 0.025
 wR factor = 0.065
Data-to-parameter ratio = 22.2For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.Trichloro(η^5 -1,2,3-trimethylcyclopentadienyl)-
titanium(IV)The crystal structure of the title compound, $[\text{Ti}(\text{Me}_3\text{Cp})\text{Cl}_3]$, where Me_3Cp is 1,2,3-trimethylcyclopentadienyl (C_8H_{11}), was determined as part of our investigation of monocyclopentadienyltitanium(IV) trihalides. The title compound exhibits a three-legged piano-stool geometry. The Ti—ring distance is 2.0120 (7) Å.

Comment

The structure of the title compound, (I), a catalyst precursor for olefin polymerization, has been analysed.

Fig. 1 shows the η^5 coordination of the trimethyl-substituted cyclopentadienyl ring to the TiCl_3 moiety, which is akin to the structures of similar half-sandwich $\text{Cp}'\text{TiCl}_3$ complexes (Cp' is a substituted cyclopentadienyl ligand).The geometry around the Ti atom can be described as a three-legged piano-stool configuration, where the plane defined by the three Cl atoms and the plane of the cyclopentadienyl ring are almost parallel, with an angle of 0.77 (7) $^\circ$ between them. The distance between the Ti atom and the centre of the ring is longer than in $[\text{Ti}(\text{MeCp})\text{Cl}_3]$ [2.0042 (15) Å] and shorter than in $[\text{Ti}(\text{Me}_4\text{Cp})\text{Cl}_3]$ [2.0151 (17) Å] or $[\text{Ti}(\text{Me}_5\text{Cp})\text{Cl}_3]$ [2.0211 (7) Å] (Kirschbaum & Giolando, 1991; Pevec, 2003).Two of the methyl C atoms, C6 and C7, are displaced slightly out of the ring plane, away from the TiCl_3 fragment (Table 1). The third C atom is displaced much less, with an angle of 0.34 (10) $^\circ$ between the ring plane and the C5—C8 bond. This fact can be attributed to the steric hindrance caused by the Cl atoms, where the $\text{Cl}2 \cdots \text{C}8$ distance [3.608 (2) Å] is longer than the $\text{Cl}2 \cdots \text{C}7$ and $\text{Cl}3 \cdots \text{C}6$ distances [3.295 (2) and 3.340 (2) Å, respectively].

Experimental

The title compound was prepared according to the trimethylsilylation procedure of Cardoso *et al.* (1980). (1,2,3-Trimethylcyclopentadienyl)trimethylsilane, obtained from 1,2,3-trimethylcyclopentadienyllithium (Broussier *et al.*, 1997) and chlorotrimethylsilane, was reacted with titanium tetrachloride (2% excess) in toluene. After evaporation of volatiles, the resulting sticky brown product was purified by repeated vacuum sublimation, yielding 42% of a red-

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orange solid. Suitable crystals of (I) were obtained by sublimation at 10^{-3} Pa, at a temperature of 353 K. Spectroscopic analysis: ^1H NMR (CDCl_3 , δ , p.p.m.): 6.72 (s, 2H), 2.40 (s, 6H), 2.33 (s, 3H); ^{13}C NMR (CDCl_3 , δ , p.p.m.): 139.6, 137.2, 123.5, 16.2, 13.3; ^{49}Ti NMR (CDCl_3 , δ , p.p.m.): -225.1 (half-width 36.9 Hz); ^{47}Ti NMR (CDCl_3 , δ , p.p.m.): -491.4 (half-width 84.6 Hz). Elemental analysis, calculated for $\text{C}_8\text{H}_{11}\text{Cl}_3\text{Ti}$: C 36.75, H 4.24, Cl 40.68%; found: C 36.28, H 4.21, Cl 40.47%.

Crystal data

$[\text{Ti}(\text{C}_8\text{H}_{11})\text{Cl}_3]$	$Z = 2$
$M_r = 261.42$	$D_x = 1.594 \text{ Mg m}^{-3}$
Triclinic, $P\bar{1}$	Mo $K\alpha$ radiation
$a = 6.6790$ (2) Å	Cell parameters from 2345 reflections
$b = 8.3360$ (3) Å	$\theta = 1-27.5^\circ$
$c = 10.7120$ (4) Å	$\mu = 1.47 \text{ mm}^{-1}$
$\alpha = 82.9070$ (19)°	$T = 150$ (2) K
$\beta = 83.340$ (2)°	Prism, red
$\gamma = 67.368$ (2)°	$0.25 \times 0.20 \times 0.20 \text{ mm}$
$V = 544.74$ (3) Å ³	

Data collection

Nonius KappaCCD area-detector diffractometer	2291 reflections with $I > 2\sigma(I)$
φ and ω scans	$R_{\text{int}} = 0.024$
Absorption correction: none	$\theta_{\text{max}} = 27.6^\circ$
8461 measured reflections	$h = -8 \rightarrow 8$
2488 independent reflections	$k = -10 \rightarrow 10$
	$l = -13 \rightarrow 13$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0307P)^2 + 0.2233P]$
$R[F^2 > 2\sigma(F^2)] = 0.025$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.065$	$(\Delta/\sigma)_{\text{max}} = 0.001$
$S = 1.06$	$\Delta\rho_{\text{max}} = 0.30 \text{ e \AA}^{-3}$
2488 reflections	$\Delta\rho_{\text{min}} = -0.34 \text{ e \AA}^{-3}$
112 parameters	
H-atom parameters constrained	

Table 1

Selected geometric parameters (Å, °).

Ti1—Cg1	2.0120 (7)	Ti1—Cl2	2.2369 (4)
Ti1—Cl1	2.2410 (5)	Ti1—Cl3	2.2412 (4)
Cg1—Ti1—Cl2	113.93 (3)	Cg1—Ti1—Cl3	115.38 (2)
Cl1—Ti1—Cl2	102.78 (2)	Cl1—Ti1—Cl3	103.30 (2)
Cg1—Ti1—Cl1	116.41 (3)	Cl2—Ti1—Cl3	103.31 (2)
P_r —C3—C6	3.59 (10)	P_r —C4—C7	3.18 (10)
P_r —C5—C8	0.34 (10)		

Cg1 is the centroid of the cyclopentadienyl ring and P_r is the ring plane defined by atoms C1–C5.

All H atoms were positioned geometrically and refined riding on their parent C atoms, with C–H = 0.93 Å and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ for

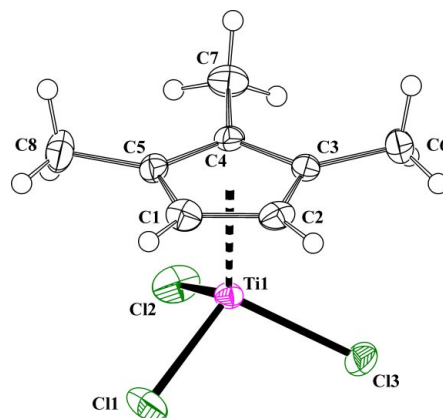


Figure 1

The structure of (I), showing 30% probability displacement ellipsoids and the atom-numbering scheme.

cyclopentadienyl H atoms, and C–H = 0.96 Å and $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$ for methyl H atoms.

Data collection: *COLLECT* (Nonius, 1998) and *DENZO* (Otwinowski & Minor, 1997); cell refinement: *COLLECT* and *DENZO*; data reduction: *COLLECT* and *DENZO*; program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1994); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 2003); software used to prepare material for publication: *SHELXL97*.

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